



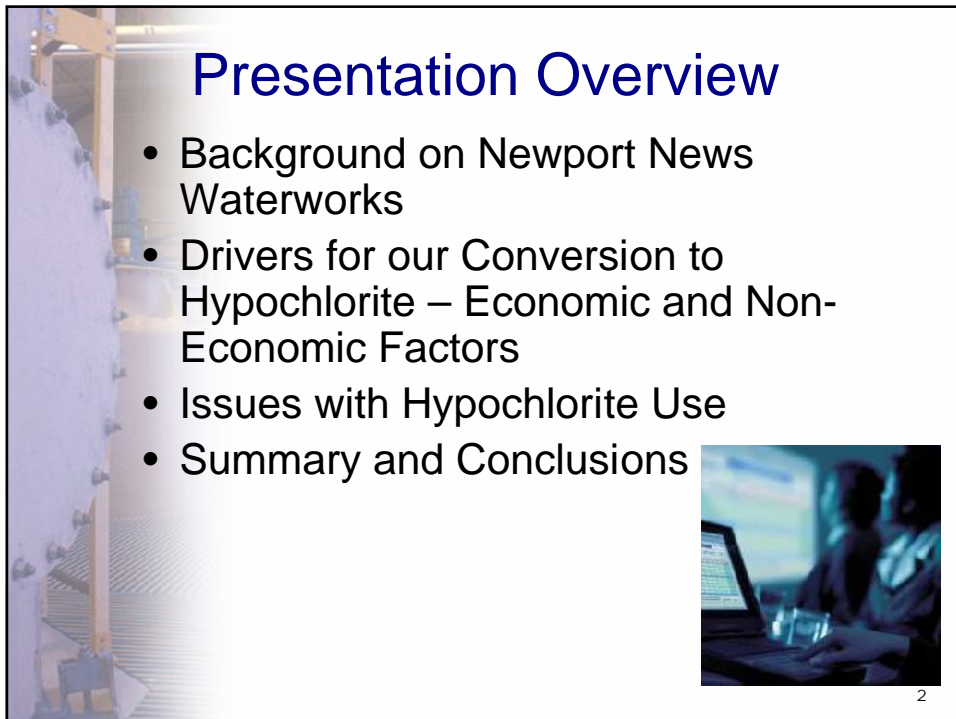
## Chlorine Gas: Balancing Public Health and Security Webcast Case Study- Conversion to Hypochlorite



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**Chief of Facilities Engineering**  
**Newport News Waterworks**




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## Presentation Overview

- Background on Newport News Waterworks
- Drivers for our Conversion to Hypochlorite – Economic and Non-Economic Factors
- Issues with Hypochlorite Use
- Summary and Conclusions



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## Newport News Waterworks

- Over 400,000 customers
- Service area includes Cities of Newport News, Hampton, Poquoson and Counties of York and portions of James City
- Average daily demand ~ 45 MGD



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## Newport News Waterworks

- Two water treatment plants
  - Harwood's Mill WTP (1989) – 48 MGD capacity
    - Liquid Chlorine Feed system using evaporators
    - Planned conversion to hypochlorite in next 3-5 years during planned plant upgrade project
  - Lee Hall WTP (2005) – 60 MGD capacity
    - Plant designed in late 1990's; construction began October 1999
    - On-site bulk storage sodium hypochlorite feed system

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## Drivers for Our Conversion to Hypo

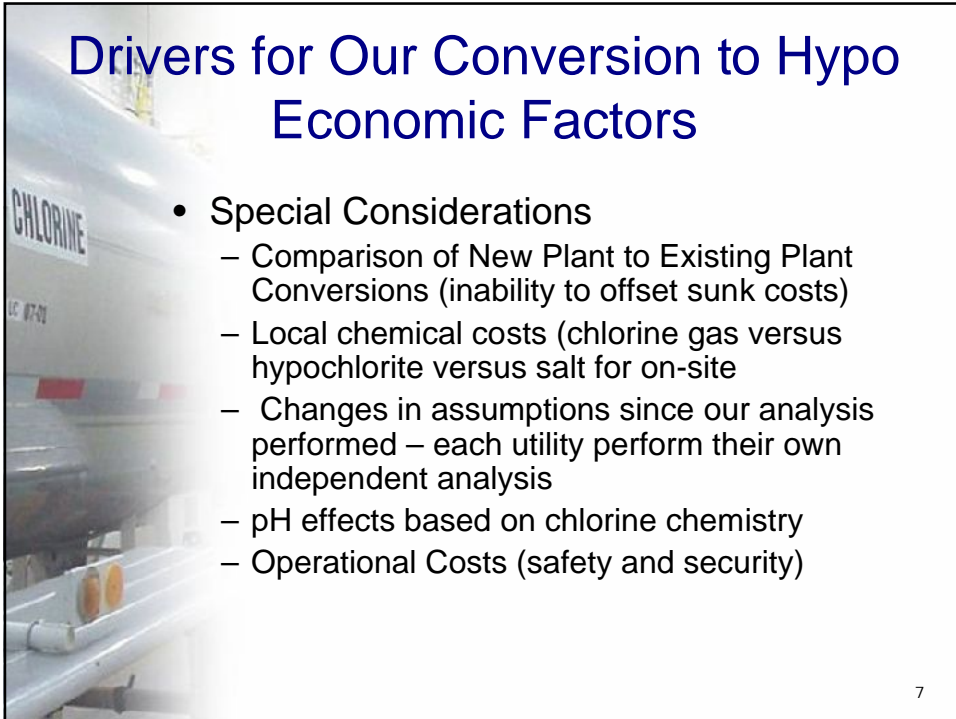
- Detailed Alternatives Analysis conducted during design
  - Evaluated several options:
    - Liquid chlorine (evaporators)
    - Gaseous chlorine (pressure/vacuum and all vacuum) – multiple cylinders connected
    - Bulk sodium hypochlorite
    - On-site generation of sodium hypochlorite
  - Developed listing of non-economic and economic factors



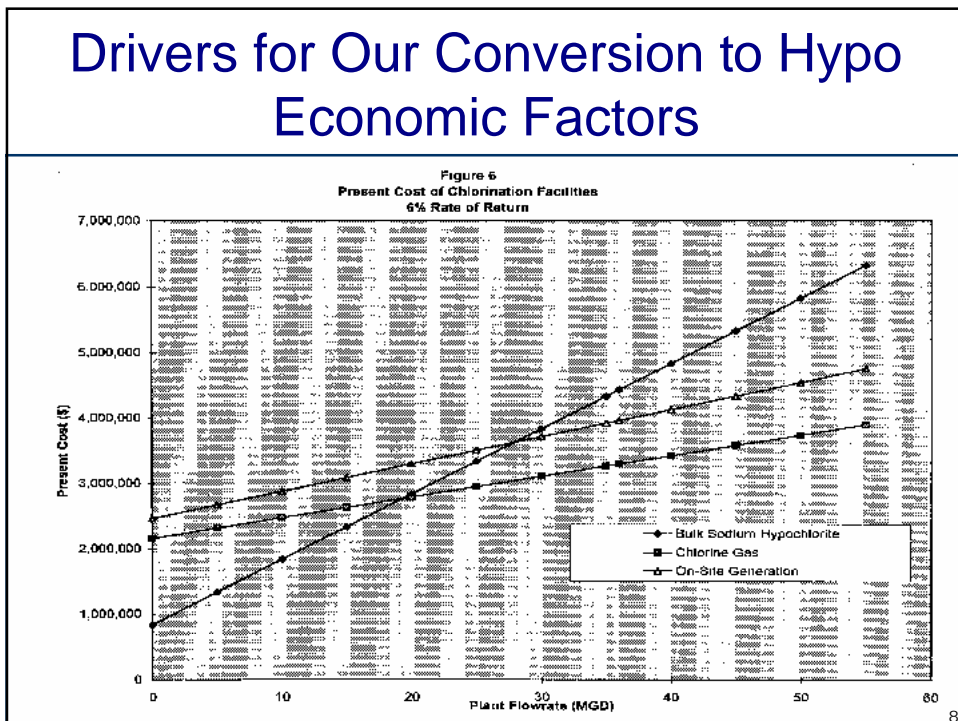
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## Drivers for Our Conversion to Hypo Economic Factors

- Special Considerations
  - Comparison of New Plant to Existing Plant Conversions (inability to offset sunk costs)
  - Local chemical costs (chlorine gas versus hypochlorite versus salt for on-site)
  - Changes in assumptions since our analysis performed – each utility perform their own independent analysis
  - pH effects based on chlorine chemistry
  - Operational Costs (safety and security)



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## Drivers for Our Conversion to Hypo

### Economic Factors

- Breakpoint analysis showed:
  - Bulk hypochlorite most economical below 20 MGD
  - Break-even cost between bulk hypo and on-site generation at approximately 28 MGD
- Since study (1997) local gaseous chlorine costs have risen by ~ 100%; local hypochlorite costs risen by ~ 33% in same period
- Transportation Costs negligible during study
- “Neighborhood” Trends



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## Drivers for Our Conversion to Hypo

### Non-Economic

- Key Non-economic Factors
  - Safety for Employees and General Public
  - Regulatory Requirements (Risk Management Plan)
  - Mitigation of Potential Accidents
  - Operational Issues
  - Security Requirements (Post 9/11 – DHS Chemical Facility Anti-Terrorism Standards)

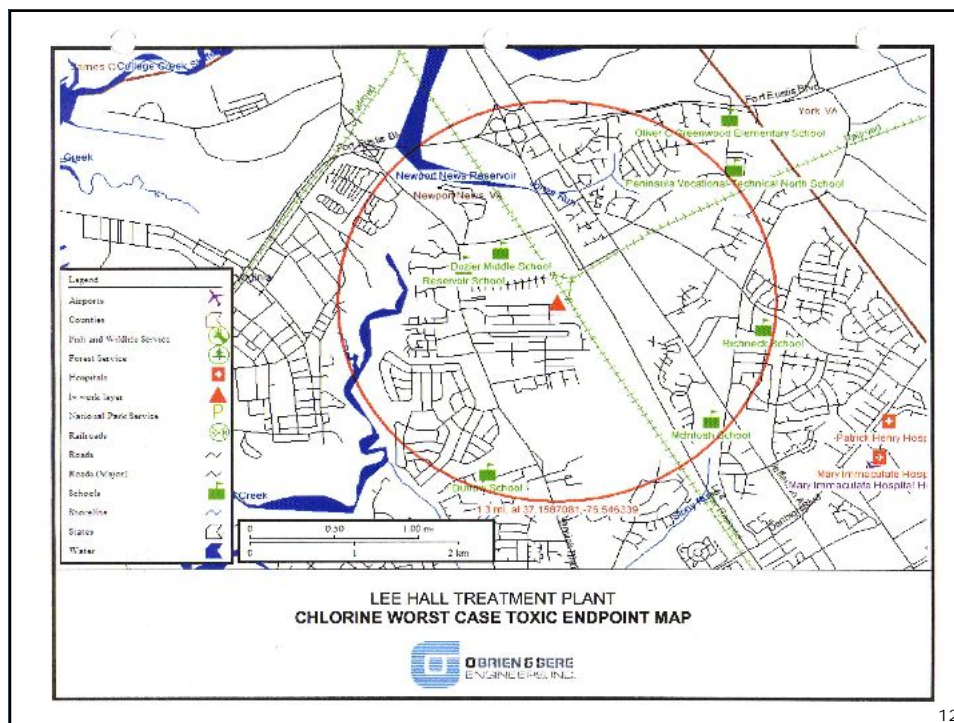
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Balancing Public Health and Security

ALTERNATIVE	ADVANTAGES	DISADVANTAGES
Liquid Chlorine System	<ul style="list-style-type: none"> <li>Requires relatively few ton containers on-line to provide the design dosage.</li> <li>Comparable to the Harwood's Mill system</li> </ul>	<ul style="list-style-type: none"> <li>System requires handling of liquid chlorine under pressure.</li> <li>System requires greater maintenance due to the need for evaporators</li> <li>Requires a Risk Management Plan per EPA and OSHA regulations</li> </ul>
Chlorine Gas (Press/Vacuum)	<ul style="list-style-type: none"> <li>Eliminates the evaporators used in liquid chlorine</li> </ul>	<ul style="list-style-type: none"> <li>Requires the use of a manifold cylinders and shutoff valves to restrict a leak to one ton container only.</li> <li>Requires a Risk Management Plan per EPA and OSHA regulations</li> </ul>
Chlorine Gas (All-Vacuum)	<ul style="list-style-type: none"> <li>Offers the greatest safety in that all pressure piping is eliminated.</li> </ul>	<ul style="list-style-type: none"> <li>Requires handling a number of vacuum regulators during a changeout of cylinders.</li> <li>Requires maintenance of many different gas filters.</li> <li>Requires a Risk Management Plan per EPA and OSHA regulations</li> </ul>
Sodium Hypochlorite (Bulk Delivery and Storage)	<ul style="list-style-type: none"> <li>Uses conventional liquid chemical handling procedures.</li> </ul>	<ul style="list-style-type: none"> <li>The 12.5% concentration of sodium hypochlorite degrades and can cause problems with valves and gas binding of pumps.</li> </ul>
Sodium Hypochlorite (On-Site Generation)	<ul style="list-style-type: none"> <li>Safest alternative requires handling salt and 0.8% solution of sodium hypochlorite.</li> </ul>	<ul style="list-style-type: none"> <li>The equipment necessary to generate the sodium hypochlorite makes this alternative have higher maintenance requirements.</li> </ul>

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## Drivers for Our Conversion to Hypo

- Bottom line driver for our utility was safety of staff and neighbors
- Break-even costs were reasonable and manageable
- Logistics of implementation can be overcome
- Currently all neighboring utilities and commercial businesses using chlorine have or are in the process of converting to hypochlorite



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## Issues with Hypochlorite Use

- Vent, vent, vent!!!
- Proper equipment selection, layout, construction orientation is critical
- Degradation of hypochlorite from original strength – storage requirements
- Materials of Construction of Bulk Storage Tanks and Gaskets
- Safety concerns



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## Issues with Hypochlorite Use



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## Issues with Hypochlorite Use



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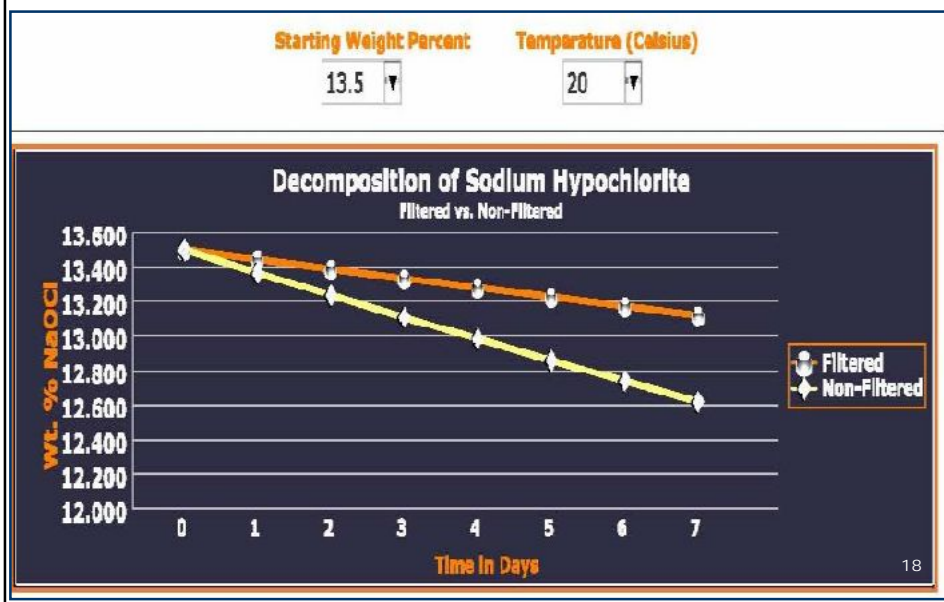


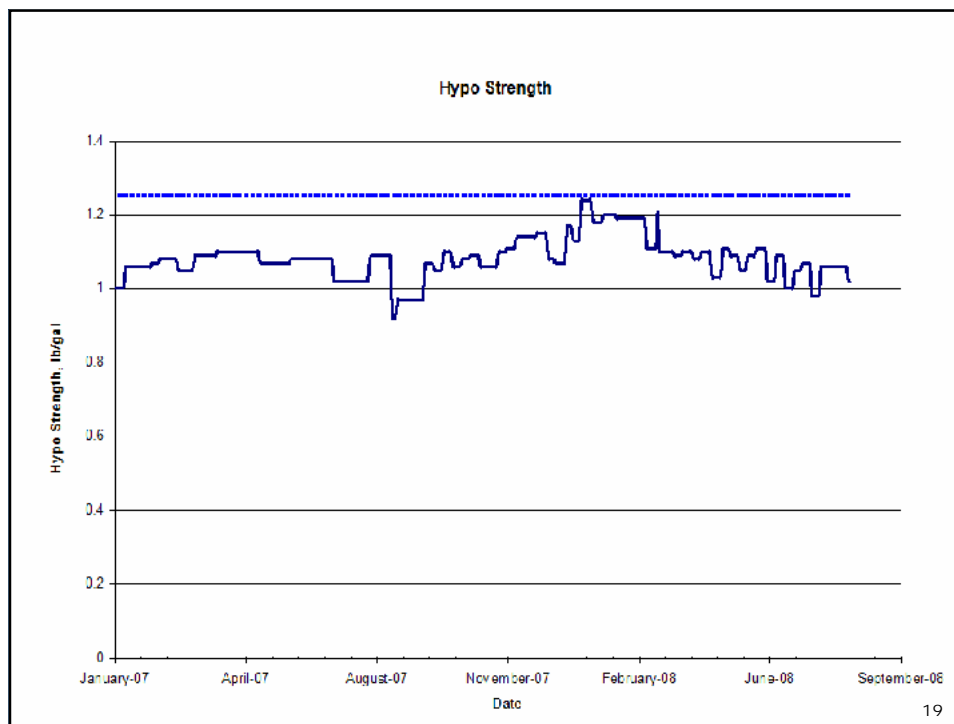
## Issues with Hypochlorite Use



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## Issues with Hypochlorite Use

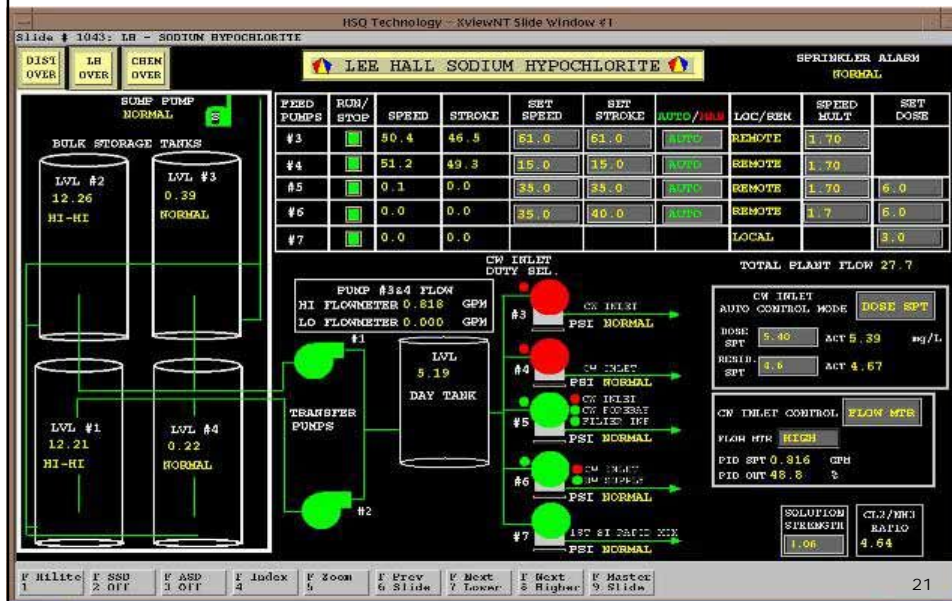




## Issues with Hypochlorite Use



## Issues with Hypochlorite Use



## Summary and Conclusions

- Each utility must develop their own listing of economic and non-economic factors and evaluate them systematically; include public input
- Hypochlorite is not without issues – handling, storage, feeding, etc.
- Local factors and regulatory requirements may drive the decision

## Ask the Experts



Bill Desing



Paul Kinshella



Scott Dewhirst



Bill Bellamy

E-mail your question with your first name, city  
and state to [online@awwa.org](mailto:online@awwa.org).

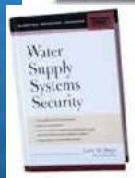
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